SYNTHESIS, CHARACTERIZATION AND ADSORPTION STUDIES OF DEEP-EUTECTIC SOLVENT MOLECULAR IMPRINTED POLYMER FOR THE REMOVAL OF BISPHENOL A

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ABSTRACT

Molecularly imprinted polymer (MIP) is a class of powerful materials with promising selective molecule recognition abilities. However, the conventional MIPs have a relatively low binding capacity and selectivity. To improve this characteristic of MIP, the addition of deep eutectic solvent (DES) as a functional monomer in the polymerization process was studied to synthesize potential MIP. The study focuses on the synthesis, characterization, and adsorption behavior of deep eutectic solvent molecularly imprinted polymer (DES-MIP), molecular imprinted polymer (MIP), and hybrid deep eutectic solvent molecular imprinted membrane (HDES-MIP) for the selective removal of bisphenol A (BPA) from aqueous medium. The potential of DES in MIP was studied by synthesizing DES-MIP using DES functional monomer. DES was synthesized by mixing choline chloride (ChCl) and methacrylic acid (MAA) by ratio of 1:2. DES-MIP was then compared with conventional MIP synthesized by MAA as a functional monomer. Both DES-MIP and MIP were prepared by the free radical polymerization process via a bulk polymerization method. Prepared DES-MIP was then hybridized with cellulose acetate (CA) to obtain the HDES-MIP membrane. The characterization results of X-ray diffraction (XRD), Fourier transform infrared spectrometry (FTIR), scanning electron microscopy (SEM), thermogravimetric analysis (TGA), and differential scanning calorimeter (DSC) confirmed the characteristics of synthesized DES-MIP, MIP and HDES-MIP. The DES-MIP has a higher selectivity for BPA (17.72 mg/g) than competing analogs, including bisphenol AP (9.78 mg/g), 2-naphthol (8.25), and 4-tertiary butyl-phenol (6.98 mg/g). The optimization parameters include adsorption pH, adsorption kinetics, adsorption isothermal, and thermodynamic studies were carried out. DES-MIP, MIP and HDES-MIP followed Pseudo-second order and Langmuir isothermal models. According to reusability tests, the DES-MIP can be recycled five times without losing significant adsorption capacity. As a result, the addition of DES in the polymerization process improved the physical and chemical properties and enhanced the recognition capacity of MIP, thus affecting the adsorption behavior.



ABSTRAK

Polimer pencetakan molekul (MIP) merupakan sejenis kelas bahan yang menakjubkan dengan jaminan kebolehan pengecaman terhadap molekul tertentu. Tetapi, MIP konvensional mempunyai kapasiti pengikat dan selektiviti yang rendah secara relatif. Untuk menambah baik ciri-ciri MIP, penambahan pelarut eutektik dalaman (DES) sebagai monomer berfungsi dalam proses pempolimeran telah dikaji untuk menghasilkan MIP yang baik. Kajian ini memfokuskan kepada penghasilan, pencirian, dan sifat penjerapan oleh pelarut eutektik dalaman polimer pencetakan (DES-MIP), polimer pencetakan molekul (MIP), dan pelarut eutektik hibrid dalaman membran cetakan molekul (HDES-MIP) sebagai penyingkiran khusus terhadap bisfenol A (BPA) daripada medium akueus. Potensi DES dalam MIP telah dikaji dengan menghasilkan DES-MIP dengan menggunakan kumpulan berfungsi DES. DES telah dihasilkan melalui campuran kolin klorida (ChCl) dan asid metakrilik (MAA) dengan nisbah 1:2. DES-MIP dibandingkan dengan MIP yang konvensional (dihasilkan dengan MAA sebagai monomer berfungsi). Kedua-dua DES-MIP dan MIP dihasilkan oleh proses pempolimeran radikal bebas melalui kaedah pempolimeran pukal. DES-MIP yang disediakan kemudian digabungkan dengan selulosa asetat (CA) untuk memperoleh membran HDES-MIP. Hasil pencirian oleh alat pembelauan sinar X (XRD), Spektroskopi inframerah fourier transformasi (FTIR), mikroskop electron pengimbas (SEM), thermal gravimetric analyzer (TGA), dan differential scanning calorimeter (DSC) telah mengesahkan ciri-ciri DES-MIP, MIP dan HDES-MIP yang dihasilkan. DES-MIP mempunyai pengkhususan yang tinggi terhadap BPA (17.72 mg/g) berbanding yang lain, seperti bisfenol AP (9.78 mg/g), 2-naftol (8.25), and 4tertiari butil-fenol (6.98 mg/g). Pengoptimuman parameter termasuk kajian penjerapan pH, penjerapan kinetik, penjerapan isotherma dan termodinamik telah dijalankan. DES-MIP, MIP dan HDES-MIP mengikuti model tertib pseudo kedua dan tertib isoterma Langmuir. Menurut kepada ujian kebolehgunaan semula, DES-MIP boleh diguna semula selama lima kali tanpa kehilangan kapasiti penjerapan yang banyak. Penambahan DES dalam proses pempolimeran menambah baik sifat-sifat fizikal dan kimianya dan meningkatkan kapasiti pengecaman MIP yang akan mempengaruhi sifat penjerapannya.

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LIST OF SYMBOLS AND ABBREVIATIONS

b Langmuir constant B_t Boyd parameter _ C_0 Concentration of the solute in the initial solution (mg/L)_ C_i Initial concentration of solution (mg/ L) _ C_f Final concentration of solution (mg /L) _ C_e Equilibrium concentration of BPA (mg /L) _ C_t Equilibrium concentration of solution (mg /L) _ Pseudo-first order rate constant (min¹) k_1 k_2 Pseudo-second order rate constant ($g mg^{-1}min^{-1}$) _ Distribution coefficients K_d Freundlich constant for sorption capacity (mg/g) K_F Freundlich constant for intensity п Amount of the BPA adsorbed at equilibrium (mg/g) q_e Amount of the BPA adsorbed at any time (mg/g) q_t Binding capacity of BPA (mg/g) Q_e Universal gas constant (kJ mol⁻¹ K⁻¹) R R_L Separation factor T Temperature (K) Time (min) t Half adsorption time (min) $T_{1/2}$ VVolume of solution (L) Mass of adsorbent particles (g) W Bisphenol A BPA _ BPO Benzoyl peroxide Choline chloride ChCl CEC Capillary electro chromatography _ DES Deep eutectic solvent _

DSC	-	Differential scanning calorimeter
DESs	-	Deep eutectic solvents
DES-MIP	-	Deep eutectic solvent molecular imprinted polymer
EDGMA	-	Ethylene glycerol dimethacrylate
EDC	-	Endocrine disturbing chemical
FRP	-	Free radical polymerization
FTIR	-	Fourier transform infrared spectroscopy
HBA	-	Hydrogen bond acceptor
HBD	-	Hydrogen bond donor
HPLC	-	High performance liquid chromatography
MAA	-	Methacrylic acid
MIPs	-	Molecular imprinted polymers
MIP	-	Molecular imprinted polymer
NIP	-	Non imprinted polymer
SPE	-	Solid phase extraction
PVC	-	Poly vinyl chloride
TGA	-	Thermogravimetric analysis
TRIM	-	Trimethylolpropane trimethacrylate
XRD	-	X-ray diffraction



LIST OF PUBLICATION

Syed Asim Hussain Shah, Mohamad Sharifah, Md Saleh Noorashikin, Yih Shiuan Beh, Rahim Yani Nurul, Miskam Mazidatulakmam and Asman Saliza*, A Review of Molecular Imprinting Polymer for Separation of Bisphenol-A and its Analogues: Synthesis and Application. *Current Analytical Chemistry* 2022; *18.* (Scopus/WOS)

CHAPTER 1

INTRODUCTION

1.1 Background of study

Molecularly imprinted polymers (MIPs) have revolutionized and attracted attention as tailor-made adsorbents for specific recognition of target molecules (Hu *et al.*, 2021). MIPs are synthetic polymers with biomimetic functionalities. MIPs are usually obtained through the polymerization of functional monomers around the target molecule (template) in a complementary fashion, bringing about an exceptional three-dimensional network (Pratiwi *et al.*, 2018). MIPs are less expensive than other purification techniques, e.g., coagulation (Eslami *et al.*, 2020), biological treatment (Saravanan *et al.*, 2021), catalytic oxidations (Wang *et al.*, 2021), ozonation solvent extraction (Yang *et al.*, 2020), and adsorption distillation (Liu *et al.*, 2021).

MIPs are based on molecular recognition, which occurs naturally. The interaction between antibodies and antigens is one example. MIPs are precise for selective identification and binding of a given guest molecule (Huang *et al.*, 2014). The extraordinary molecular interaction between monomer receptor and desired analyte is a result of MIPs well-designed molecular recognition. The synthesis of MIPs usually starts with the pre-assembly of monomers in the presence of template in a suitable soluble medium. The monomer is arranged around the template in a fixed position during the process, which is followed by copolymerization of crosslinker and initiation (Zhao *et al.*, 2019). The template molecule is eliminated after polymerization leaving behind tri-dimensional structure with size, shape, and functional group complementary to the template molecule.

MIPs showed applications as separation materials for the analysis of various compounds, including drugs (Walshe *et al.*, 1997), pesticides (Yang *et al.*, 2006) an

amino acids (Dinu & Apetrei, 2022). The MIP membrane is a specific detection technology that has been increasing in the last few years. Several articles have been published on the preparation of MIP membranes, showing specific permeability and separation for template/ligands such as cholesterol (Ciardelli *et al.*, 2006), nucleotides, and various drugs (Trotta *et al.*, 2005). The transport properties and applications of MIP membranes in sensor technology have also been investigated (Algieri *et al.*, 2014).

Besides all the advantages, conventional MIPs are still facing the problem of low binding capacity and low selectivity (Arabi *et al.*, 2020). Deep eutectic solvents (DESs) are emerging solvents gradually introduced as functional monomers in molecular imprinting (Jablonský *et al.*, 2020). DESs are comprised of mixture of two or more constituents that have a low melting point as compared to their components (Wang *et al.*, 2018). One component behaved as a hydrogen bond donor (HBD) while the other one acted as a hydrogen bond acceptor (HBA). These components are held together with each other by either hydrogen bond or van der Waals interaction. DESs possess excellent properties, such as functional capacity, low cost, and low toxicity (Fu *et al.*, 2017). DESs enhance the aggregate of the imprinting site marking the selective adsorption capacity of the surface.



Additionally, DESs have distinct properties, including easy preparation, low vapour pressure, and favourable biodegradability (Qin *et al.*, 2019). The selectivity and adsorption capacity of target material is successfully enhanced by the introduction of DESs on the surface of molecular imprinted polymer (MIP) and this system is called deep eutectic solvent molecularly imprinted polymer (DES-MIP). DES-MIP showed good stability, excellent compatibility, improved selectivity, reusability, better imprinting factor, quick binding kinetics, and greater adsorption capacity. The excellent improvement of MIPs upon modification with DES is due to the controlled structure and homogeneity of the binding site (Wang *et al.*, 2018).

Bisphenol A (BPA) is an emerging organic, synthetic, and chemical intermediate used in certain plastics and epoxy resins (Fan *et al.*, 2021). BPA is the most widespread endocrine disruptor which caused chronic health conditions related to the reproductive system (Shamhari *et al.*, 2021), metabolic function, nervous system, immune function, growth, and development of progeny (Ma *et al.*, 2019). The general population may directly or indirectly exposed to BPA through food, water, air, cosmetics, and thermal. Eventually, these products are biodegraded by sunlight and

converted into microplastics. Due to its non-biodegradability and resistance to chemical degradation, it is essential to remove BPA from the environment (Huang *et al.*, 2017).

In this study profoundly specific and proficient DES based on choline chloridemethacrylic acid (ChCl-MAA) as a functional monomer was developed. The obtained DES was used as a functional monomer to produce (DES-MIP). A conventional MIP based on MAA monomer was also produced. The prepared DES-MIP was hybridized with cellulose acetate (CA) to obtain the hybrid DES-MIP membrane (HDES-MIP).

1.2 Problem statement

MIPs are synthetic polymers that contain artificial receptor sites for a specific target template. Many researchers have investigated MIPs for BPA (Alnaimat *et al.*, 2019; Ardekani *et al.*, 2020; Poliwoda *et al.*, 2016; Y. Wang *et al.*, 2019). Most of these studies used conventional functional monomers such as methacrylic acid (MAA), acrylamide (AAM) and 4-vinyl pyridine (4-PV). MIPs obtained through conventional monomers still face limitations. Poor selectivity, irregular shape, and slow mass transfer lead to poor binding capacity for the target template (Tian *et al.*, 2018).



The release of a wide range of chemical pollutants with potential toxicity into water is an ecological threat to humans. BPA is one of the major chemical manufactured worldwide due to the growing market of polycarbonates and epoxy resins (Huang *et al.*, 2012). The estimated use of BPA was 7.7 million metric tons in 2015 and is projected to reach 10.6 million metric tons by 2022. Commonly, BPA is used to synthesize polycarbonates, phenolics, and epoxies. Industrial wastes, effluents and household products are some of the common sources of pollution of BPA in the environment (Kwiatkowska *et al.*, 2017). It is negligibly released into the environment mixed up with drinking water and food by contacting polycarbonates and resins, glassware, food cans, baby bottles, and storage buckets.

BPA is responsible for causing tumors in humans, even taking a minor concentration every day. It is extremely important to remove BPA as the exposure leads to severe health and environmental effects (Corrales *et al.*, 2015). BPA has been banned in the production of baby bottles along with any other food contact items for children under the age of three. The European Chemical Agency (ECHA) has

classified BPA as a chemical of very high concern. Recent rules have placed even more restrictions on the use of BPA that's why food manufacturers are steadily looking for an alternative to eliminate BPA from their goods (Wassenaar et al., 2021). The European Commission approved a regulation in 2018 that restricted the use of BPA as a packing material (Zhang et al., 2020). The Specific Migration Limit (SML) per kilogram of food was reduced from 0.6 mg to 0.05 mg. Per body weight, 4 ug/kg was set as the new Tolerable Daily Intake (t-TDI).

A new generation of functional DES monomers has some unique characteristics compared to traditional monomers, such as the creation of inclusion complexes by host-guest interaction DES forms a complex with the target analyte through various types of intermolecular interactions such as Van der Waals force, hydrophobic interactions, electrostatic affinity, dipole-dipole and hydrogen bond interaction during the imprinting Phases (Grecco et al., 2021). Thus, DES is the alternative approach that eliminates the drawbacks faced by using traditional JKU TUN AMINAH monomers.

1.3 **Objectives of study**



The main objective of this study is to improve the selectivity and binding capacity of molecular imprinted polymer (MIP) by introducing a choline chloride/methacrylic base functional monomer. Meanwhile, the specific objectives are as follows:

- To synthesize and characterize the deep eutectic solvent molecular imprinted i.D polymer (DES-MIP) for the removal of bisphenol A (BPA).
- ii. To prepare and characterize the hybrid DES-MIP membrane (HDES-MIP) for the removal of BPA.
- iii. To study the optimum parameters for DES-MIP and hybrid DES-MIP membrane (HDES-MIP) for the removal of BPA.

1.4 Significance of study

The conventional MIPs faced several problems, including low selectivity and low binding capacity. Application of new functional monomer for MIP enhanced selectivity and binding capacity for BPA. DES is an excellent choice as a functional

monomer for the preparation of DES-MIP due to less toxic, low cost and easy preparation made. Molecular imprinting technology (MIT) was revolutionized after the utilization of DES, which proved to be an effective source to overcome the drawbacks of traditional molecular imprinting. Factors like incomplete template removal, irregular shape and low binding capacity can be improved by the introduction of DES for the MIP synthesis. The proposed method for the synthesis of DES-MIP's remarkably enhanced binding capacity. Consequently, excellent template recognition, as well as adsorption capacity have been achieved.

Additionally, the wide range of polarity of DESs makes it possible to construct analytical platforms for several compounds for which it has previously been impossible to do so due to solubility issues. For example, DESs have recently been employed as extraction solvents for the pre-concentration of pesticide residues in food and environmental samples, followed by instrumental analysis. Finally, DESs are nontoxic and eco-friendly, making them desirable for use as green media in the creation of devices for the pharmaceutical, agrochemical, and food sectors. The advantages of the existence of high content functional groups in DES monomers provide excellent interaction with template molecules which tend to enhance the selectivity and affinity of DES-MIP. The effective extraction ability of DES-MIP is due to the presence of natural process action hydrogen bonding, hydrophobic nature, and electrostatic interaction (Li *et al.*, 2015)



1.5 Scope of study

In order to achieve the research objectives, the scopes of study have been determined are:

- i. Synthesis of DES-MIP and DES-NIP by preparing a new functional monomer which is combination of choline chloride-methacrylic acid (ChCl-MAA) as a DES compound.
- Synthesis of the conventional MIP and NIP as control by using methacrylic acid (MAA) as functional monomer.
- iii. Synthesis and characterization of hybrid deep eutectic solvent molecular imprinted polymer membrane (HDES-MIP).

- iv. Evaluation of interaction between functional groups of DES, MIPs, NIPs and HDES-MIP was done by using Fourier transform infrared spectrometer (FTIR).
- v. Evaluation of morphology of synthesized MIPs, NIPs and HDES-MIP was analyzed by using a scanning electron microscope (SEM).
- vi. Evaluation of crystalline properties of MIPs and NIPs were carried out by using X-ray diffraction (XRD) analysis.
- vii. Determination of heating behavior of MIPs and NIPs was carried out by Differential scanning calorimetry (DSC).
- viii. Evaluation of thermal stability with respect to temperature was carried out by Thermogravimetric analysis (TGA) for MIPs and NIPs
- ix. Evaluation of optimum parameters including pH, kinetic, isothermal and thermodynamic of MIPs and HDES-MIP was carried out by a batch adsorption experiment using UV-vis spectrophotometry.
- x. Determination of selectivity of MIPs and NIPs was carried out by comparing the adsorption of structurally similar molecules.
- xi. Evaluation of reusability of MIPs was carried out by performing five adsorption cycles.



CHAPTER 2

LITERATURE REVIEW

2.1 Molecularly imprinted polymer (MIPs)

2.1.1 Overview of MIPs

Molecular imprinting is an outstanding approach for the synthesis of polymers with high selectivity and specificity for the target molecule, which involves the attachment of polymerizable functional monomers around the template molecules (Bates *et al.*, 2017). Molecularly imprinted polymers (MIPs) have a specific ability to determine and separate specific molecules (target template) from other molecules that contain similar structures. Due to specific recognition, easy preparation and low-cost MIPs find their application in the areas of purification, separation, biosensing, catalysis, drug delivery, and degradation (Chen *et al.*, 2016).Three major approaches are used to synthesize MIPs (i) covalent, (ii) non-covalent, and (iii) semi-covalent.



The noncovalent imprinting approach is frequently used to prepare MIPs due to its flexibility and ease of handling. The noncovalent interactions involve Van der Waals forces, hydrogen bonding and π - π interactions (Zhang *et al.*, 2021). Most common functional monomer for this type of MIP is methacrylic acid (MAA), which interacts through hydrogen bonding. The imprint molecules interact with the MIPs during the imprinting procedure and the rebinding process via non-covalent interactions. Simple and economical experimental procedures of noncovalent approach allow a wide range of functional monomers to interact with almost every kind of template available commercially (Martín-Esteban, 2016). A major limitation of this approach is the formation of complexes with multiple template-monomer stoichiometries. Such complexes may have binding sites with different affinities. However, the simplicity of operation, kinetics of binding removal and versatility of the



Figure 2.1: Schematic preparation of MIP by non-covalent approach (Gao *et al.*, 2020)



The covalent approach was first introduced by Wulff and coworkers (Wulff, 1977). It is the second most popular approach used for the synthesis of MIPs. The covalent approach indicates that the cleavage of the covalent reversible bond occurs to remove MIPs from the resulting matrix, which is reformed upon rebinding of the analyte (Parisi et al., 2019). The covalent bonds must be cleaved for the template to be extracted via chemical interaction, resulting in well-defined binding cavities with complementary steric and functional topography to the target molecule. The exact stoichiometry of the template-monomer complexes allows the preparation of polymers with binding groups exclusively located in the imprinted cavities, decreasing the possibility of non-specific interactions. There are some limitations associated with the covalent approach for MIPs. The application of the covalent approach is restricted to a limited number of functional monomers and templates such as alcohols, amines, aldehydes, ketones and carboxylic acids (Marfà et al., 2021). Moreover, templatemonomer complex formation before polymerization is attributed to higher effort for template cleavage after MIP synthesis. The repeated use and slow rebinding kinetics are due to the need to establish the covalent bond for target recognition

The covalently BPA-imprinted polymer was prepared using BPA as a template, dimethacrylate as the monomer, ethylene glycol dimethacrylate (EGDMA) as the crosslinker, and 2,2'-azobis (isobutyronitrile) as the initiator. The polymerization was polymerized by UV initiation by using chloroform as a solvent. The BPA was separated from the resulting polymer by hydrolysis of the ester bonds with aqueous sodium hydroxide, generating the carboxylic acid residues in the polymer. The ester bonds of the obtained polymer were hydrolyzed to generate carboxyl group-based binding sites. The resulting polymer exhibited a high affinity for BPA. Table 2.1 shows the advantages and disadvantages of the three approaches used in the synthesis of MIPs (Ikegami *et al.*, 2004).

Imprinting type	Benefits	Limitations	References
Covalent	Provide homogeneous binding sites with a definite shape. Polymerization ensures the maximization of the number of specific binding sites.	The removal of template and rebinding is very difficult.	(Hasanah <i>et al.</i> , 2021; Yi <i>et al.</i> , 2013)
Semi-covalent	Random distribution and association of functional monomers to template molecules.	There may be template bleeding when hydrolysis is unable to remove the template.	(Hasanah <i>et al</i> ., 2021)
Non-covalent	Experimental simplicity. Easy removal of the template under mild conditions. Fast template binding and release kinetics.	The weak average affinity of the binding site. Formation of nonselective binding sites because of random incorporation of	(Torres-Cartas et al., 2020)

Table 2.1: Summary	of different approaches	s used for the synt	hesis of MIPs
2	11	2	

The semi-covalent approach is a protocol for producing MIPs that combines the covalent and noncovalent approaches. This alternative approach can provide a higher imprinting factor and a faster adsorption value. (Zhang *et al.*, 2020) applied a semi covalent approach for MIP to fabricate the mesoporous fluorescent molecularly imprinted sensor for detecting BPA from food samples. To achieve rapid identification of BPA, an imprinting precursor1-(benzofuran-2-yl)-2-propylaminopentane (BPAP) was formulated via thermally reversible isocyanate bonding that worked as an alternative template molecule for BPA detection.

2.2 Synthesis of MIPs

MIPs are commonly synthesized by free radical polymerization (FRP). The other components involved in the synthesis process are functional monomer, template, crosslinker and initiator.

2.2.1 Free radical polymerization (FRP)

FRP is widely used in the industrial preparation of polymers. FRP-prepared materials have distinct functionalities. As a result, approximately 50% of the world's polymer production is typically carried out using this method (Marfà *et al.*, 2021). MIPs are often synthesized by FRP which is the most important method for their production. In the presence of a suitable solvent, the reaction is carried out under the normal condition of temperature and pressure.

Typically, the synthesis technique is carried out in bulk or solution at mild reaction temperatures lower than 80 °C. This range of temperature is quite favorable for wide range of functional groups and templates (Kouki *et al.*, 2022). Typically, the polymerization reaction happens quite quickly. A popular azo-initiator used in this reaction is azo N-N'-bis isobutyronitrile (AIBN), which can initiate a reaction thermally or chemically. Previous studies demonstrated that photo-initiated polymerization at low temperature decreases the kinetic energy of the prepolymerization complex which increased the stability and allowing greater binding capacity and specificity than thermal initiated polymerization which requires temperatures higher than 40 °C.

As shown in Figure 2.2, free FRP consists of three steps: (i) initiation, (ii) propagation, and (iii) termination. Initiation is the first step in which the formation of free radicals occurs by the decomposition of an initiator under heating. The second major step is the propagation, where the addition of a monomer to a growing macro radical contributes to the growth of the chain. The final step is termination, which takes place in two ways. First is the recombination of two macro radicals and secondly, the disproportionation that gives the double bonds C=C and C-H at the end of the chain of polymer (Datta & Włoch, 2019).



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VITA

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